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## A nimesulide intermediate

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### Abstract

The X-ray structure of the title compound, C<sub>13</sub>H<sub>13</sub>NO<sub>3</sub>S, has been determined. The two phenyl rings make a dihedral angle of 73.5 (1)°, indicating the non-planarity of the molecule. The orientation of the methanesulfonamide moiety to the phenyl ring is given by torsion angle C13—S—N—C12 = 69.2 (2)°. The molecules are linked into pairs by N—H···O hydrogen bonding over an inversion centre.

### Comment

The structure of the title compound was examined as part of an extensive study of Non-steroidal anti-inflammatory drugs (Ravikumar, 1993, 1994). We report here the crystal structure of 2-phenoxyethanesulfonamide, (I), the precursor used in the synthesis of Nimesulide, (II), an anti-inflammatory drug (Moore & Harrington, 1974; Swingle & Moore, 1984; Ward & Brodgen, 1988). A view of the molecule with atom numbering is shown in Fig. 1. The S=O distances [1.424 (2)–1.438 (2) Å] are within the range observed for other sulfonamide drugs (Chatterjee *et al.*, 1981). The structure of Nimesulide has been reported (Dupont *et al.*, 1995). Bond distances and angles are similar to those observed in Nimesulide except C8—C9—C10 angle, 120.5 (2)°. The widening of the corresponding angle 123.1 (3)° observed in Nimesulide may be attributed to the effect of the nitro group substituted at C9. The dihedral angle between the two phenyl rings is 73.5 (1)° [74.7 (1)° in Nimesulide (II)] indicating that the molecule is twisted. In the crystal, the molecules exist as N—H···O hydrogen bond dimers around inversion centres. The amino N atom and atom O2 of the sulfonamide group participate in the hydrogen bond formation [N···O2(−x, −y + 1, −z) 2.949 Å, H0A···O2 2.153 Å and N—H0A···O2 147.1°].

### Experimental

The title compound was provided by NATCO Laboratory, Hyderabad, India, and recrystallized from benzene by slow evaporation.

### Refinement

H atoms were located from difference Fourier map; they were positioned geometrically and included as riding atoms with fixed isotropic displacement parameter in the structure-factor calculations.

### Computing details

Data collection: P3 Diffractometer Program (Siemens, 1991); cell refinement: *SHELXTL-Plus* (Sheldrick, 1991); data reduction: *SHELXTL-Plus*; program(s) used to solve structure: *SHELXTL-Plus*; program(s) used to refine structure: *SHELXL93* (Sheldrick, 1993); molecular graphics: *SHELXTL-Plus*; software used to prepare material for publication: *SHELXL93*.

## 2-phenoxymethanesulfonanilide

### Crystal data

$C_{13}H_{13}NO_3S$	$\gamma = 102.78 (1)^\circ$
$M_r = 263.30$	$V = 627.0 (1) \text{ \AA}^3$
Triclinic, $P\bar{1}$	$Z = 2$
$a = 8.146 (1) \text{ \AA}$	Mo $K\alpha$
$b = 8.927 (1) \text{ \AA}$	$\mu = 0.26 \text{ mm}^{-1}$
$c = 9.290 (1) \text{ \AA}$	$T = 293 (2) \text{ K}$
$\alpha = 101.14 (1)^\circ$	$0.16 \times 0.16 \times 0.16 \text{ mm}$
$\beta = 100.95 (1)^\circ$	

### Data collection

Siemens R3m/V diffractometer	$R_{\text{int}} = 0.018$
Absorption correction: none	2 standard reflections
2669 measured reflections	every 98 reflections
2481 independent reflections	intensity decay: $\leq 1\%$
2077 reflections with $I > 2\sigma(I)$	

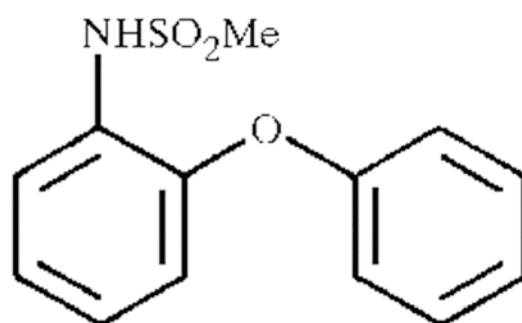
### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.040$	164 parameters
$wR(F^2) = 0.116$	H-atom parameters constrained
$S = 1.04$	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
2481 reflections	$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$

## References

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Scheme 1



**supplementary materials**

2-phenoxymethanesulfonanilide

Crystal data

$C_{13}H_{13}NO_3S$	$Z = 2$
$M_r = 263.30$	$F_{000} = 276$
Triclinic, $P\bar{1}$	$D_x = 1.395 \text{ Mg m}^{-3}$
$a = 8.146 (1) \text{ \AA}$	Mo $K\alpha$ radiation
$b = 8.927 (1) \text{ \AA}$	$\lambda = 0.71073 \text{ \AA}$
$c = 9.290 (1) \text{ \AA}$	Cell parameters from 25 reflections
$\alpha = 101.14 (1)^\circ$	$\theta = 6\text{--}15^\circ$
$\beta = 100.95 (1)^\circ$	$\mu = 0.26 \text{ mm}^{-1}$
$\gamma = 102.78 (1)^\circ$	$T = 293 (2) \text{ K}$
$V = 627.0 (1) \text{ \AA}^3$	Transparent cube, colorless
	$0.16 \times 0.16 \times 0.16 \text{ mm}$

Data collection

Siemens R3m/V diffractometer	$R_{\text{int}} = 0.018$
Radiation source: fine-focus sealed tube	$\theta_{\text{max}} = 26.1^\circ$
Monochromator: graphite	$\theta_{\text{min}} = 2.3^\circ$
$T = 293(2) \text{ K}$	$h = 0 \rightarrow 10$
$\omega/2\theta$ scans	$k = -11 \rightarrow 10$
Absorption correction: none	$l = -11 \rightarrow 11$
2669 measured reflections	2 standard reflections
2481 independent reflections	every 98 reflections
2077 reflections with $I > 2\sigma(I)$	intensity decay: $\leq 1\%$

Refinement

Refinement on $F^2$	Secondary atom site location: difference Fourier map
Least-squares matrix: full	Hydrogen site location: inferred from neighbouring sites
$R[F^2 > 2\sigma(F^2)] = 0.040$	H-atom parameters constrained
$wR(F^2) = 0.116$	Calculated $w = 1/[\sigma^2(F_o^2) + (0.062P)^2 + 0.2251P]$
$S = 1.04$	where $P = (F_o^2 + 2F_c^2)/3$ ?
2481 reflections	$\Delta\rho_{\text{max}} = 0.20 \text{ e \AA}^{-3}$
164 parameters	$\Delta\rho_{\text{min}} = -0.16 \text{ e \AA}^{-3}$
Primary atom site location: structure-invariant direct methods	Extinction correction: SHELXL93, $F_c^* = kFc[1 + 0.001xFc^2\lambda^3/\sin(2\theta)]^{-1/4}$
	Extinction coefficient: 0.160 (10)

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )

x	y	z	$U_{\text{iso}}^*/U_{\text{eq}}$
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## supplementary materials

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S	0.05903 (7)	0.58633 (6)	0.26066 (6)	0.0377 (2)
N	0.1093 (2)	0.4278 (2)	0.1847 (2)	0.0371 (4)
H0A	0.1102	0.4075	0.0862	0.080*
O1	-0.1320 (2)	0.1599 (2)	0.1693 (2)	0.0486 (4)
O2	0.0399 (2)	0.6717 (2)	0.1448 (2)	0.0503 (4)
O3	0.1836 (2)	0.6620 (2)	0.4003 (2)	0.0531 (4)
C1	-0.3990 (3)	0.0271 (3)	0.2121 (3)	0.0479 (5)
H1A	-0.4031	0.1257	0.2728	0.080*
C2	-0.5316 (3)	-0.1088 (3)	0.1868 (3)	0.0560 (6)
H2B	-0.6275	-0.1039	0.2321	0.080*
C3	-0.5256 (3)	-0.2503 (3)	0.1005 (3)	0.0534 (6)
H3B	-0.6190	-0.3438	0.0832	0.080*
C4	-0.3878 (3)	-0.2559 (3)	0.0376 (3)	0.0580 (6)
H4A	-0.3828	-0.3539	-0.0234	0.080*
C5	-0.2539 (3)	-0.1200 (3)	0.0609 (3)	0.0542 (6)
H5A	-0.1568	-0.1224	0.0168	0.080*
C6	-0.2620 (3)	0.0187 (2)	0.1485 (2)	0.0386 (5)
C7	0.0271 (3)	0.1787 (2)	0.2627 (2)	0.0358 (4)
C8	0.0686 (3)	0.0706 (3)	0.3433 (3)	0.0455 (5)
H8A	-0.0158	-0.0268	0.3344	0.080*
C9	0.2340 (3)	0.1031 (3)	0.4339 (3)	0.0490 (6)
H9A	0.2625	0.0294	0.4908	0.080*
C10	0.3574 (3)	0.2410 (3)	0.4439 (3)	0.0505 (6)
H10A	0.4715	0.2634	0.5082	0.080*
C11	0.3165 (3)	0.3482 (3)	0.3621 (2)	0.0439 (5)
H11A	0.4036	0.4429	0.3681	0.080*
C12	0.1513 (3)	0.3192 (2)	0.2725 (2)	0.0345 (4)
C13	-0.1435 (3)	0.5254 (3)	0.3009 (3)	0.0560 (6)
H13A	-0.1768	0.6163	0.3467	0.080*
H13B	-0.2285	0.4683	0.2089	0.080*
H13C	-0.1359	0.4577	0.3693	0.080*

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
S	0.0433 (3)	0.0326 (3)	0.0355 (3)	0.0083 (2)	0.0071 (2)	0.0088 (2)
N	0.0437 (10)	0.0346 (9)	0.0334 (8)	0.0083 (7)	0.0110 (7)	0.0103 (7)
O1	0.0388 (8)	0.0389 (8)	0.0626 (10)	0.0023 (6)	-0.0017 (7)	0.0228 (7)
O2	0.0670 (10)	0.0396 (8)	0.0474 (9)	0.0152 (7)	0.0118 (8)	0.0187 (7)
O3	0.0588 (10)	0.0418 (8)	0.0453 (9)	0.0078 (7)	-0.0021 (8)	0.0009 (7)
C1	0.0479 (13)	0.0438 (12)	0.0505 (13)	0.0135 (10)	0.0123 (10)	0.0063 (10)
C2	0.0447 (13)	0.0577 (14)	0.066 (2)	0.0088 (11)	0.0212 (12)	0.0147 (12)
C3	0.0472 (13)	0.0451 (12)	0.0599 (14)	-0.0005 (10)	0.0084 (11)	0.0138 (11)
C4	0.065 (2)	0.0378 (12)	0.067 (2)	0.0090 (11)	0.0196 (13)	0.0039 (11)
C5	0.0512 (13)	0.0463 (12)	0.069 (2)	0.0123 (10)	0.0268 (12)	0.0112 (11)
C6	0.0361 (10)	0.0346 (10)	0.0431 (11)	0.0056 (8)	0.0035 (9)	0.0151 (8)
C7	0.0366 (10)	0.0368 (10)	0.0347 (10)	0.0097 (8)	0.0084 (8)	0.0105 (8)
C8	0.0490 (12)	0.0411 (11)	0.0490 (12)	0.0118 (10)	0.0111 (10)	0.0184 (10)

C9	0.0556 (14)	0.0497 (13)	0.0460 (12)	0.0239 (11)	0.0059 (10)	0.0169 (10)
C10	0.0443 (12)	0.0568 (14)	0.0469 (12)	0.0202 (10)	0.0006 (10)	0.0077 (11)
C11	0.0386 (11)	0.0432 (11)	0.0447 (12)	0.0086 (9)	0.0074 (9)	0.0038 (9)
C12	0.0390 (10)	0.0341 (10)	0.0313 (9)	0.0113 (8)	0.0100 (8)	0.0069 (8)
C13	0.0521 (14)	0.065 (2)	0.0568 (14)	0.0189 (12)	0.0214 (12)	0.0168 (12)

*Geometric parameters (Å, °)*

S—O3	1.424 (2)	C3—C4	1.366 (4)
S—O2	1.438 (2)	C4—C5	1.389 (3)
S—N	1.627 (2)	C5—C6	1.366 (3)
S—C13	1.753 (2)	C7—C8	1.390 (3)
N—C12	1.438 (2)	C7—C12	1.402 (3)
O1—C7	1.371 (2)	C8—C9	1.382 (3)
O1—C6	1.409 (2)	C9—C10	1.381 (4)
C1—C6	1.368 (3)	C10—C11	1.390 (3)
C1—C2	1.381 (3)	C11—C12	1.381 (3)
C2—C3	1.376 (3)		
O3—S—O2	118.96 (10)	C5—C6—C1	121.7 (2)
O3—S—N	108.05 (10)	C5—C6—O1	120.3 (2)
O2—S—N	105.55 (9)	C1—C6—O1	117.9 (2)
O3—S—C13	107.87 (12)	O1—C7—C8	124.8 (2)
O2—S—C13	108.39 (11)	O1—C7—C12	114.6 (2)
N—S—C13	107.51 (11)	C8—C7—C12	120.6 (2)
C12—N—S	120.98 (13)	C9—C8—C7	119.3 (2)
C7—O1—C6	118.1 (2)	C10—C9—C8	120.5 (2)
C6—C1—C2	118.7 (2)	C9—C10—C11	120.2 (2)
C3—C2—C1	120.8 (2)	C12—C11—C10	120.3 (2)
C4—C3—C2	119.5 (2)	C11—C12—C7	119.1 (2)
C3—C4—C5	120.4 (2)	C11—C12—N	120.6 (2)
C6—C5—C4	118.9 (2)	C7—C12—N	120.3 (2)
O3—S—N—C12	47.0 (2)	C6—O1—C7—C12	177.4 (2)
O2—S—N—C12	175.27 (15)	O1—C7—C8—C9	180.0 (2)
C13—S—N—C12	-69.2 (2)	C12—C7—C8—C9	0.3 (3)
C6—C1—C2—C3	0.6 (4)	C7—C8—C9—C10	-0.4 (4)
C1—C2—C3—C4	-0.9 (4)	C8—C9—C10—C11	-0.4 (4)
C2—C3—C4—C5	0.3 (4)	C9—C10—C11—C12	1.3 (3)
C3—C4—C5—C6	0.5 (4)	C10—C11—C12—C7	-1.5 (3)
C4—C5—C6—C1	-0.7 (4)	C10—C11—C12—N	-178.5 (2)
C4—C5—C6—O1	-177.4 (2)	O1—C7—C12—C11	-179.1 (2)
C2—C1—C6—C5	0.2 (3)	C8—C7—C12—C11	0.7 (3)
C2—C1—C6—O1	176.9 (2)	O1—C7—C12—N	-2.0 (3)
C7—O1—C6—C5	-74.1 (3)	C8—C7—C12—N	177.8 (2)
C7—O1—C6—C1	109.1 (2)	S—N—C12—C11	-82.6 (2)
C6—O1—C7—C8	-2.3 (3)	S—N—C12—C7	100.4 (2)